

### 18.4 Acid-Base Chemistry: Theory

#### Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities (correct)?  
 (A) pressure, concentration (B) pressure, pH  
 (C) molar fraction (D) molar fraction, volume
2. Which of the following is CORRECT? (A) concentration of ions increases as pH increases (B) concentration of ions increases as pH decreases



3. In a 0.1 M acetic acid solution, the concentration of acetate ions is 0.01 M. The pH of the solution is approximately (A) 2.0 (B) 3.0 (C) 4.0 (D) 5.0



4. In a 0.1 M acetic acid solution, the concentration of acetate ions is 0.01 M. The pH of the solution is approximately (A) 2.0 (B) 3.0 (C) 4.0 (D) 5.0



5. Which of the following is CORRECT? (A) concentration of ions increases as pH increases (B) concentration of ions increases as pH decreases



6. A solution of acetic acid and its conjugate base is prepared with a pH of 4.0. The concentration of acetic acid is 0.1 M. The concentration of acetate ions is (A) 0.01 M (B) 0.02 M (C) 0.05 M (D) 0.1 M



7. In a 0.1 M acetic acid solution, the concentration of acetate ions is 0.01 M. The pH of the solution is approximately (A) 2.0 (B) 3.0 (C) 4.0 (D) 5.0



# Multiple Choice Stoichiometry Test

**E Durkheim**



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