
ACS ANALYTICAL CHEMISTRY STUDY GUIDE

- k_f : $k_f = k_f \cdot a$
 - EDTA titrations: 1. before eq EDTA binds to metal 1:1
2. At eq we assume that all metal ions can bind to EDTA and then use k_f
- Potentiometry: an analytical method in which an electric potential difference (avoltage) of a cell is measured
 - Reducing agent: Donates electrons and becomes oxidized.
 - oxidizing agent: Accepts electrons and becomes reduced.
 - coulometry: science of measuring change and relating it to analyte
 - cathode: the electrode at which reduction occurs
 - Anode: the electrode at which oxidation occurs
 - E_{cell} : $E_{cell} = E_{red} - E_{ox}$
 - Faraday's Equation: $E = E - (0.0592/n) \log(a/b)$
 - ||: anode always written on left of ||
 - amperometry: elec techniques where the cell current is measured where pot diff between the indicator and reference electrode is controlled
 - Clark electrode: Electrode for oxygen
 - Biosensors: Detecting devices that can detect specific substances in cells
 - amp chromo det: point, size, or charge
 - limiting current: the diff between current and steady state max
 - Anodic vol: Very sensitive for analyzing metals into ppb
 - fluorescence: used to measure singlets, atoms like rigid aromatic rings, high sensitivity
 - phosphorescence: involves emission of light trip- singlet, much lower than fluoro
 - Transmittance: $A = -\log(T)$ or $2 - \log(T\%)$
 - atomic absorption: lamp-flame-mono-detector
 - atomic emission: flame-mono-detector
 - AA hollow cathode lamp: used to form very narrow band lights
 - AA atomizer: Heat sample to very high temps to break all chemical bonds, hot enough to

Physical Chemistry Acs Exam Study Guide

Thomas A. Holme, Kristen Murphy



Physical Chemistry Acs Exam Study Guide:

ACS Physical Chemistry Sterling Test Prep,2025-04-26 ASC Physical Chemistry bestsellerThousands of students use Sterling Test Prep study aids to achieve high test scores Comprehensive review of physical chemistry topics tested on ACS covering the knowledge areas necessary to answer test questions Understanding key concepts extracting and analyzing information and distinguishing between similar answer choices are more effective than mere memorization This book helps you master physical chemistry college content and develop the ability to apply your knowledge on the exam This text is clearly presented and systematically organized to provide targeted ACS Physical Chemistry preparation Learn the scientific foundations and essential concepts needed to master the material and answer exam questions These review chapters teach important physical chemistry principles and relationships and how they apply to questions From the foundations of physical and chemical properties of matter to complex mechanisms of atomic particles you will develop a better understanding of chemical phenomena on microscopic and macroscopic levels This book is an essential overview of chemical kinetics reaction mechanisms thermodynamics and quantum chemistry concepts Learn about intra and intermolecular interactions properties of gases liquids and solids chemical and phase equilibria atomic theory and electronic structure Experienced chemistry instructors analyzed the test content and developed this study material that builds knowledge and skills crucial for success Our test preparation experts systematized the content to match the current ACS exam for targeted and effective learning

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