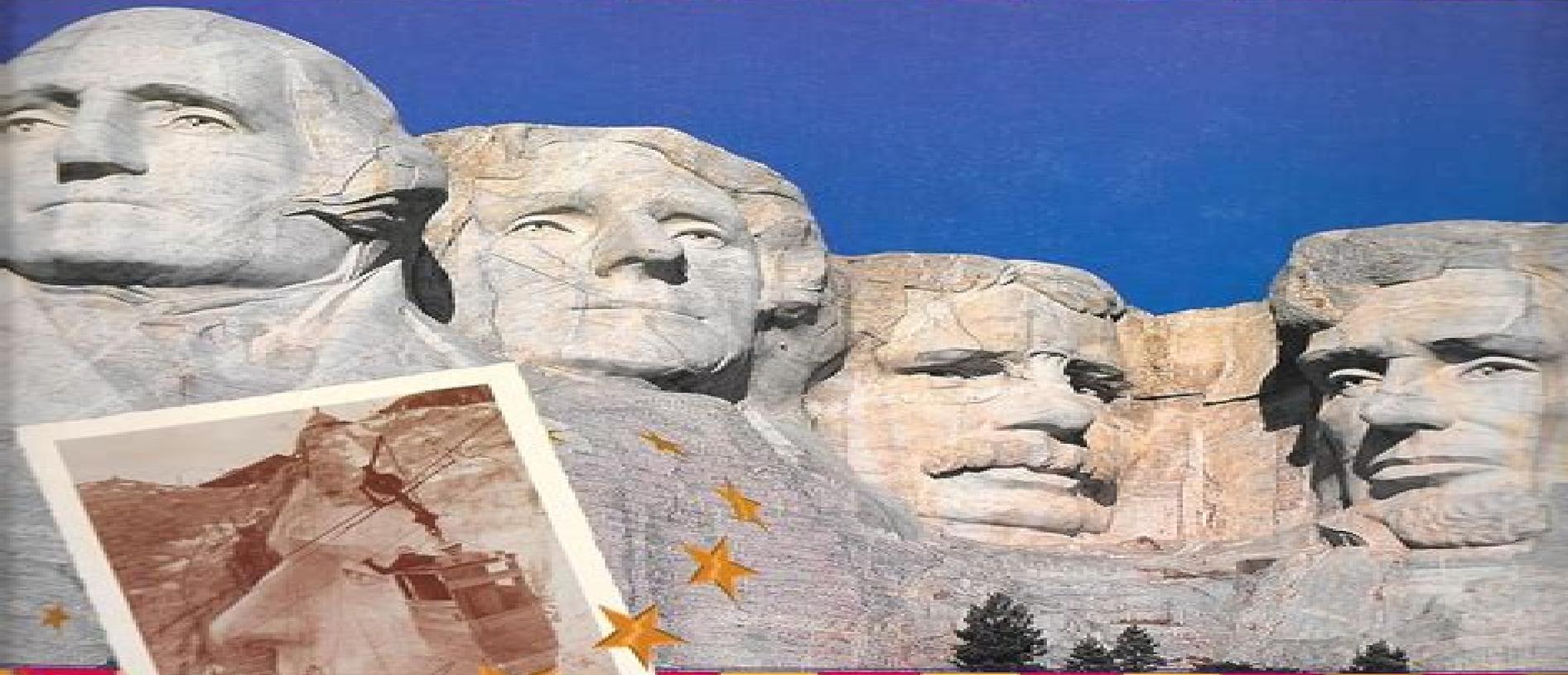


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 oxidation $\text{CO} + \text{CO}_2 + 2\text{e}^-$ reduction $\text{Fe} + \text{Fe}_2\text{O}_3 + 3\text{e}^-$ $24 \text{ Ag} + \text{Ni}^{3+} \rightarrow \text{Ag} + \text{Ni}$ oxidation reduction $25 \text{ Cu} + \text{Pb}^{2+} \rightarrow \text{Cu}^{2+} + \text{Pb}$ oxidation reduction 26 O_2

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 $5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + \text{I}_2$ $\text{b} \text{ O}_5 + 5\text{C} \rightarrow \text{O}_2 + 5\text{CO}_2$

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