

### 18.4 Acid-Base Chemistry: Theory

#### Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance  
 (B) volume, molarity, pH, amount of substance  
 (C) volume, molarity, pH, concentration of ions  
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A)  $n = 1000$ ,  $m = 1000$

(B)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$   
 (C)  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

2. An acid-base reaction represents a proton transfer from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.  
 (B) The equilibrium constant is less than 1.  
 (C) The equilibrium constant is equal to 1.  
 (D) The equilibrium constant is greater than 1.

(A)  $n = 1000$ ,  $m = 1000$

(B)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

3. An acid-base reaction involves the transfer of a proton from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.  
 (B) The equilibrium constant is less than 1.  
 (C) The equilibrium constant is equal to 1.  
 (D) The equilibrium constant is greater than 1.

(A)  $n = 1000$ ,  $m = 1000$

4. Which of the following is not a property of a strong acid? (correct)

(A)  $n = 1000$ ,  $m = 1000$

(B)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

5. A solution of a weak acid and its conjugate base is prepared. The pH of the solution is 4.0. The acid dissociation constant is  $1.0 \times 10^{-4}$ . What is the ratio of the concentration of the acid to the concentration of the conjugate base? (correct)

(A)  $n = 1000$ ,  $m = 1000$

(B)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

6. An acid-base reaction involves the transfer of a proton from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.  
 (B) The equilibrium constant is less than 1.  
 (C) The equilibrium constant is equal to 1.  
 (D) The equilibrium constant is greater than 1.

(A)  $n = 1000$ ,  $m = 1000$

# Multiple Choice Stoichiometry Test

**M Mark**



**Multiple Choice Stoichiometry Test:**

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