

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A) 10^{-14} (B) 10^{-7}

2. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

3. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

4. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

5. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

6. Which of the following is correct? (correct)

(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

7. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

8. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
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(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

9. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

10. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

Multiple Choice Stoichiometry Test

Liying Dong



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