

A quick guide to Harvard referencing

Why reference at all?

Referencing is an important part of academic writing, and it is something that you may find yourself doing later in your career as well. We reference for a lot of different reasons but at the heart of it lies a few key principles; acknowledge, authority, findability.

Acknowledge: We reference to acknowledge the sources we have used to build our own understanding of a topic. This demonstrates that these points are not necessarily our own work or ideas but that they have helped formed our own arguments. It also helps us to avoid plagiarism.

Authority: By demonstrating that we have looked at other key thinkers and researchers in an area, we add weight and authority to our argument. It shows that we have a nuanced and critical understanding of the issues and that we have based our thinking on evidence.

Findability: Lastly, a good reference will signpost your reader so that they can read and learn more about this topic. Don't underestimate this part; think about how often you use someone else's references to discover more about a certain subject.

Parts of a reference

Referencing is usually done in two parts; the in-text citation and the full reference, usually listed at the end of your essay or assignment.

In-text citation

The in-text citation gives immediate acknowledgment to the author(s) of the research or work you are referring to. It is used to indicate that a particular argument or the supporting evidence you are using has originated with someone else. It should contain the author(s) surname and date of publication.

Reference

The reference list at the end of your assignment will give the full information about the source you are referencing. You must include the full reference for every piece of work you cite. This information will help direct your reader to the original source so they can learn more about the points you are discussing, whilst acknowledging the contribution of the original author to this field of interest.

How to reference...

A book with one author

Citation order: Surname, Initial. (Year of publication) *Title of book in italics*. Edition number if relevant. Place of publication: Publisher.

Example: **In-text citation:** Botting (2008, p.19) states that...

Reference list: Botting, F. (2008) *Gothic romanced: consumption, gender and technology in contemporary fictions*. London: Routledge.

Harvard Graphics Quick Reference

J Elliott



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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton $h_a + b \rightarrow hb + a^-$ strong acids e.g. sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the K_a or K_b from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral ph at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web 1 approximately 3 g of hydrated oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution pH is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then $\text{pH} = \log 0.100 = 1.000$ pH log 0.100 = 1.000 b titrant volume 12.50 ml

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les sentiers de la perdition ou la voie de perdition au québec road to perdition est un film américain réalisé par sam mendes sorti en 2002 il est adapté de la bande dessinée du même nom écrite par max allan collins et dessinée par richard piers rayner en publiée par dc comics en 1998

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avant de devenir un film les sentiers de la perdition est un comics créé par max allan collins celui ci avoue avoir été inspiré et influencé par un classique de la bande dessinée

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