

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. metals \_\_\_\_\_
- b. metal ions \_\_\_\_\_
- c. anodes \_\_\_\_\_
- d. cathodes \_\_\_\_\_
- e. half-reaction oxidation: Zn \_\_\_\_\_
- f. half-reaction reduction: \_\_\_\_\_

- g. write overall balanced redox reaction: \_\_\_\_\_
- h. E<sub>cell</sub> \_\_\_\_\_

i. Write overall cell equation:  $\text{Zn} + \text{Cu}^{2+} \rightarrow \text{Zn}^{2+} + \text{Cu}$  (do not allow electrons to appear in overall equation) (1 mark)

\_\_\_\_\_

# Voltaic Cells Study Guide

**JE Gale**



## **Voltaic Cells Study Guide:**

## Reviewing **Voltaic Cells Study Guide**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Voltaic Cells Study Guide**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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