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web in a calorimeter the white solid $CaSO_4$ forms and the temperature of the mixture increases to 28 1 c assuming that the

calorimeter absorbs only a negligible quantity of heat and that the specific heat capacity of the solution is $4.18 \text{ J } ^\circ\text{C}^{-1} \text{ g}^{-1}$ and that the density of the final solution is $1.0 \text{ g } \text{mL}^{-1}$ calculate the enthalpy change per mole

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