

18.4 Molecular Structure: Mass

Multiple Choice:

1. The molar masses of quantities in chemical equations are based _____
 (A) percent composition (B) percent yield
 (C) molecular weight (D) stoichiometric coefficients
2. Percent of hydrogen in $\text{C}_2\text{H}_6\text{O}$ is _____ of the molecular weight.
 (A) 15.0% (B) 12.5%

3. In a gas the other reactants represent oxygen and sulfur dioxide, 2.5 g of gas was produced by the reaction, and only 0.5 g of gas, identify the percent yield of sulfur in this reaction?

4. In a gas the other reactants represent oxygen and sulfur dioxide, 2.5 g of gas was produced by the reaction, and only 0.5 g of gas, identify the percent yield of sulfur in this reaction?
 (A) 20.0% (B) 25.0% (C) 30.0% (D) 35.0% (E) 40.0%
5. In a gas the other reactants represent oxygen and sulfur dioxide, 2.5 g of gas was produced by the reaction, and only 0.5 g of gas, identify the percent yield of sulfur in this reaction?
 (A) 15.0% (B) 20.0% (C) 25.0% (D) 30.0% (E) 35.0%

6. In a gas the other reactants represent oxygen and sulfur dioxide, 2.5 g of gas was produced by the reaction, and only 0.5 g of gas, identify the percent yield of sulfur in this reaction?
 (A) 15.0% (B) 20.0% (C) 25.0% (D) 30.0% (E) 35.0%

7. Percent of hydrogen in $\text{C}_2\text{H}_6\text{O}$ is _____ of the molecular weight.
 (A) 15.0% (B) 12.5% (C) 10.0% (D) 7.5%

8. Molar mass of water and H₂ are similar, however the molar masses of these reactants below, if 2.5 g of reactants is used and 0.5 g of gas was produced, what mass of H₂ was produced?
 (A) 1.0 (B) 1.5 (C) 2.0 (D) 2.5 (E) 3.0

9. In a gas the other reactants represent oxygen and sulfur dioxide, 2.5 g of gas was produced by the reaction, and only 0.5 g of gas, identify the percent yield of sulfur in this reaction?
 (A) 15.0% (B) 20.0% (C) 25.0% (D) 30.0% (E) 35.0%

Multiple Choice Stoichiometry Test

Yicheng Fang



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