

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) volume, molarity, pH, amount of substance
2. Which of the following is CORRECT? (correct)
- (A) $\text{pH} = -\log_{10} [\text{H}^+]$
 (B) $\text{pH} = -\log_{10} [\text{H}_3\text{O}^+]$
 (C) $\text{pH} = -\log_{10} [\text{H}^+]$ and $\text{pH} = -\log_{10} [\text{H}_3\text{O}^+]$
 (D) $\text{pH} = -\log_{10} [\text{H}^+]$ and $\text{pH} = -\log_{10} [\text{H}_3\text{O}^+]$ and $\text{pH} = -\log_{10} [\text{H}^+]$ and $\text{pH} = -\log_{10} [\text{H}_3\text{O}^+]$
3. In a 0.10 mol dm⁻³ solution, the concentration of H⁺ ions is 0.001 mol dm⁻³. What is the pH of the solution? (correct)
- (A) 1.0
 (B) 1.3
 (C) 1.0
 (D) 1.3
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Ardas Ozsogomonyan



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