

18.4 Write the following in words:

18.4.1 Write in words:

1. The concentration of quantities in the following equations is written _____
 (a) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (b) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 (c) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (d) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

2. Write in words the following equations in symbols (use the following symbols):



- (a) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (b) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (c) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 (d) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (e) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (f) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

3. In a given chemical equation, the reactants are written on the left-hand side and the products are written on the right-hand side. Write the symbols for the reactants and products in the following equations:



- (a) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (b) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (c) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (d) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

4. In the following equations, write the symbols for the reactants and products in the spaces provided:



5. Write in words the following equations in symbols (use the following symbols):



- (a) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (b) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (c) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (d) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

6. Write in words the following equations in symbols (use the following symbols):



- (a) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (b) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (c) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (d) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

7. In the following equations, write the symbols for the reactants and products in the spaces provided:



Multiple Choice Stoichiometry Test

Raffaela Di Napoli

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular glow behind it.

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