

MODULE 1: ORGANIC CHEMISTRY

Organic Chemistry

- Organic compounds
- Contains carbon
 - Except CO_2 , CO_3^{2-}
- Classify according to functional group
 - Family/class
 - Similar structure and reactivity
 - Ex. CH_3OH (methanol), $\text{CH}_3\text{CH}_2\text{OH}$ (ethanol)
 - "-ol" – hydroxyl functional group

Hydrocarbons (C & H)

C-C	ALKANES	C=O	CARBONYL GROUP
C=C	ALKENES	RCOH	ALDEHYDES (CHO – Formyl)
C≡C	ALKYNES	RCOOH	CARBOXYLIC ACIDS (COOH – Carboxyl)
	AROMATIC COMPOUNDS / ARENES	CARBOXYLIC DERIVATIVES	
RX	ALKYL HALIDES	RCOOR	ESTERS
RNH ₂	AMINES	RCONH ₂	AMIDES
ROR	ETHERS	CONH ₂	CARBAMOYL
RSR	THIOETHERS (super analog of ethers)	RCOOCOR	ANHYDRIDES
RCH	ALCOHOLS	RCOX	ACYL HALIDE
RSH	THIOLS (sulfhydryl/mercapto)	RCO	ACYL
RC=OR	KETONES	CH ₃ CO	ACETYL (simplest acyl)

NOMENCLATURE

- Organic compounds may have different names
- Ex. CH_3COOH – acetic acid(semi-systematic)/ethanoic acid(systematic)/(dil.) vinegar(trivial)

IUPAC (International Uni...)

- Formulated the rules on nomenclature

Format:

- Prefix-parent-bond type-suffix
- Parent
 - Most important
 - Usually the longest chain

- Bond type
 - Single "an"
 - Double "en"
 - Triple bond "yn"
- Prefixes
 - 1 meth
 - 2 eth
 - 3 prop
 - 4 but
 - 5 pent
- Suffix & Prefix
 - Substituents
 - Suffix
 - Highest priority substituent only
 - RCOOH
 - RCOOH derivatives
 - RCHO
 - RCOR
 - ROH
 - RSH
 - RNH₂
 - C-C
 - C=C
 - C-C
 - RX, ROR, RSR – never suffix

Ex. $\text{CH}_3\text{CH}_2\text{Cl}$

✓ Ethyl chloride

✓ Chloroethane

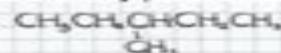
$\text{CH}_3\text{CH}_2\text{OCH}_2\text{CH}_3$

✓ Ethyl ether

✓ Ethoxyethane

Practice Set 1

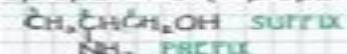
- Pentane: 3-methylpentane



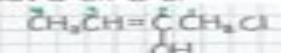
- Rule: alphabetization
3-ethyl-4-methylhexane



- 2-aminopropan-1-ol (1-propanol)



- 1-chlorobut-2-en-2-ol



- for multiple subs., add multiplier prefix
1,1,2-trimethylcyclohexane



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Lingsheng Yao



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Table of Contents Form And Function An Organic Chemistry Module Teachers Guide

1. Understanding the eBook Form And Function An Organic Chemistry Module Teachers Guide
 - The Rise of Digital Reading Form And Function An Organic Chemistry Module Teachers Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Form And Function An Organic Chemistry Module Teachers Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Form And Function An Organic Chemistry Module Teachers Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Form And Function An Organic Chemistry Module Teachers Guide
 - Personalized Recommendations
 - Form And Function An Organic Chemistry Module Teachers Guide User Reviews and Ratings

- Form And Function An Organic Chemistry Module Teachers Guide and Bestseller Lists
- 5. Accessing Form And Function An Organic Chemistry Module Teachers Guide Free and Paid eBooks
 - Form And Function An Organic Chemistry Module Teachers Guide Public Domain eBooks
 - Form And Function An Organic Chemistry Module Teachers Guide eBook Subscription Services
 - Form And Function An Organic Chemistry Module Teachers Guide Budget-Friendly Options
- 6. Navigating Form And Function An Organic Chemistry Module Teachers Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Form And Function An Organic Chemistry Module Teachers Guide Compatibility with Devices
 - Form And Function An Organic Chemistry Module Teachers Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Form And Function An Organic Chemistry Module Teachers Guide
 - Highlighting and Note-Taking Form And Function An Organic Chemistry Module Teachers Guide
 - Interactive Elements Form And Function An Organic Chemistry Module Teachers Guide
- 8. Staying Engaged with Form And Function An Organic Chemistry Module Teachers Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Form And Function An Organic Chemistry Module Teachers Guide
- 9. Balancing eBooks and Physical Books Form And Function An Organic Chemistry Module Teachers Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Form And Function An Organic Chemistry Module Teachers Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Form And Function An Organic Chemistry Module Teachers Guide
 - Setting Reading Goals Form And Function An Organic Chemistry Module Teachers Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Form And Function An Organic Chemistry Module Teachers Guide
 - Fact-Checking eBook Content of Form And Function An Organic Chemistry Module Teachers Guide
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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