

### 18.4 Acid-Base Chemistry: Theory

#### Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance  
 (B) volume, molarity, pH, amount of substance  
 (C) volume, molarity, pH, concentration of ions  
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A)  $\text{pH} = 10.0$ ,  $\text{pH} = 10.00$

(B)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$   
 (C)  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

2. An acid-base indicator has a  $\text{pK}_a$  of 4.5. In a solution of this indicator, the ratio of the concentration of the acid form to the concentration of the base form is 10:1. What is the pH of the solution? (correct)

(A)  $\text{pH} = 4.5$ ,  $\text{pH} = 4.5$ ,  $\text{pH} = 4.5$ ,  $\text{pH} = 4.5$

(B)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

3. An acid-base indicator has a  $\text{pK}_a$  of 4.5. In a solution of this indicator, the ratio of the concentration of the acid form to the concentration of the base form is 10:1. What is the pH of the solution? (correct)

(A)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

4. Which of the following is a dimensional quantity? (correct)

(A)  $\text{pH} = 10.0$ ,  $\text{pH} = 10.00$

(B)  $10 \text{ g}$ ,  $1 \text{ mol}$ ,  $1 \text{ g}$ ,  $1 \text{ mol}$ ,  $10 \text{ g}$ ,  $1 \text{ mol}$

5. A solution of an acid has a pH of 4.5. The concentration of the acid is 0.1 mol/L. What is the concentration of the base? (correct)

(A)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

(B)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

6. An acid-base indicator has a  $\text{pK}_a$  of 4.5. In a solution of this indicator, the ratio of the concentration of the acid form to the concentration of the base form is 10:1. What is the pH of the solution? (correct)

(A)  $\text{pH} = 10.0$ ,  $\text{pH} = 10.00$ ,  $\text{pH} = 10.0$ ,  $\text{pH} = 10.00$

# Multiple Choice Stoichiometry Test

**CO Houle**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red-to-white gradient semi-circle overlapping the right side of the bar.

**Multiple Choice Stoichiometry Test:**

The book delves into Multiple Choice Stoichiometry Test. Multiple Choice Stoichiometry Test is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Multiple Choice Stoichiometry Test, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
    - Chapter 1: Introduction to Multiple Choice Stoichiometry Test
    - Chapter 2: Essential Elements of Multiple Choice Stoichiometry Test
    - Chapter 3: Multiple Choice Stoichiometry Test in Everyday Life
    - Chapter 4: Multiple Choice Stoichiometry Test in Specific Contexts
    - Chapter 5: Conclusion
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  3. In chapter 2, the author will delve into the foundational concepts of Multiple Choice Stoichiometry Test. The second chapter will elucidate the essential principles that must be understood to grasp Multiple Choice Stoichiometry Test in its entirety.
  4. In chapter 3, the author will examine the practical applications of Multiple Choice Stoichiometry Test in daily life. The third chapter will showcase real-world examples of how Multiple Choice Stoichiometry Test can be effectively utilized in everyday scenarios.
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  6. In chapter 5, this book will draw a conclusion about Multiple Choice Stoichiometry Test. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Multiple Choice Stoichiometry Test.

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